

TSSMTM
Creating VCE Success

Exam Practice Guide

Unit 1

Physics

Examination 1

Key Features:

- ✓ 109 original examination style questions on all examinable topics.
- ✓ Full solutions and a marking guide to all questions.
- ✓ Written by VCE assessors who mark the real examinations.
- ✓ Excellent resource for examination practice.

Helping VCE students be the best they can be.

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SAMPLE

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AREA OF STUDY 1: HOW CAN THERMAL EFFECTS BE EXPLAINED**Key knowledge 1: THERMODYNAMIC PRINCIPLES****Question 1**

Convert the following temperatures from Celsius to Kelvin.

- i. $37^{\circ}\text{C} = \underline{\hspace{2cm}}\text{K}$
- ii. $-7^{\circ}\text{C} = \underline{\hspace{2cm}}\text{K}$
- iii. $50^{\circ}\text{C} = \underline{\hspace{2cm}}\text{K}$

3 marks

Question 2

Kelvin to Celsius

- i. $527\text{ K} = \underline{\hspace{2cm}}^{\circ}\text{C}$
- ii. $270\text{ K} = \underline{\hspace{2cm}}^{\circ}\text{C}$
- iii. $351\text{ K} = \underline{\hspace{2cm}}^{\circ}\text{C}$

3 marks

Question 3

Describe thermal equilibrium.

2 marks

Question 4

Define the Zeroth law of thermodynamics.

2 marks

Question 5

Is it true that the first law of thermodynamics is a special case of the law of conservation of energy? Explain your answer.

2 marks

Question 6

State the first law of thermodynamics.

2 marks

Question 7

While 348 kJ of work is done on a system, the system releases 543 kJ of heat. Calculate the change in internal energy.

2 marks

Question 8

If the system presented in Question 7 undergoes a second stage where it absorbs 89 kJ of heat while 53 kJ of work is done on it. Find the total change in internal energy.

2 marks